

High-temperature reactions in the ternary $\text{VO}_2\text{-Nb}_2\text{O}_5(\text{Ta}_2\text{O}_5)\text{-Fe}_2\text{O}_3$ systems

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The binary and ternary mixture oxides of the transition elements with the general composition MO_2 , e.g. CrVSbVO_6 , $\text{Cr}_2\text{V}_2\text{WO}_{10}$, $\text{Cr}_2\text{Nb}_2\text{WO}_{10}$, $\text{NiV}_2\text{Nb}_2\text{O}_{10}$, CrVNbO_6 , FeVSbO_6 , $\text{Fe}_{1-x}\text{Cr}_x\text{VSbO}_6$ show interesting electrical, magnetic and structural properties [1-5]. Some of these phases are currently being investigated as potential candidates for electrodes in the rechargeable lithium batteries [6] and as materials in the photoelectrolysis of water [7].

The investigations of solid-state reactions occurring among the oxides V_2O_4 (VO_2), Fe_2O_3 and Nb_2O_5 showed that in the system V-Fe-Nb-O the compound with the rutile-type structure is formed, possessing the formula FeVNbO_6 [1]. The research indicated that this compound was obtained in the following manner: first $\text{V}_2\text{Nb}_2\text{O}_9$ was prepared by heating either a V_2O_4 and Nb_2O_5 or V_2O_5 and NbO_2 mixture in evacuated quartz tubes at 1000°C . Then Fe_2O_3 was mixed with $\text{V}_2\text{Nb}_2\text{O}_9$ in 1:1 stoichiometric quantities and heated in vacuum at 1000°C .

The aim of the present work was, first of all, answering the question whether a compound analogous to FeVNbO_6 is formed in the system $\text{VO}_2\text{-Ta}_2\text{O}_5\text{-Fe}_2\text{O}_3$. In such a case our further aim was carrying out a primary research on the structure and the thermal properties of these compounds. The present study is to confirm whether or not solid solutions with structure $\text{FeVNb}(\text{Ta})\text{O}_6$, FeNbO_4 and/or FeTaO_4 are formed in the investigated systems. The aim was also to investigate their range of homogeneity and thermal stability.

The research was started with attempts to synthesise the compounds FeVNbO_6 and FeVTaO_6 . For this purpose, the several samples were prepared from the oxides VO_2 , Nb_2O_5 (Ta_2O_5) and Fe_2O_3 . The reacting substances were weighed in appropriate portions, homogenized by grinding in the mortar, shaped into pellets and heated, depending on the composition, in the temperature range $600\text{-}1200^\circ\text{C}$.

The kind of phases contained in the samples was identified on the base of X-ray phase analysis results (the diffractometer EMPYREAN II, PANalytical with $\text{CuK}\alpha/\text{Ni}$ radiation).

The DTA-TG investigations in the temperature range $20\text{-}1500^\circ\text{C}$ were performed by means of an SDT 2960 (TA Instruments).

Preliminary results showed that in the systems $\text{VO}_2\text{-Nb}_2\text{O}_5\text{-Fe}_2\text{O}_3$ and $\text{VO}_2\text{-Ta}_2\text{O}_5\text{-Fe}_2\text{O}_3$, the solid solutions with the orthorhombic structure FeNbO_4 as well as FeTaO_4 , are formed. The research in order to determine whether in the $\text{VO}_2\text{-Nb}_2\text{O}_5(\text{Ta}_2\text{O}_5)\text{-Fe}_2\text{O}_3$ systems also form the solid solutions with the rutile-type structure is continued.

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